



REVEAL

REVOLUTIONARY ENERGY STORAGE CYCLE
WITH CARBON FREE ALUMINIUM

Aluminium as an Energy Carrier...

... for heat and electricity in winter

Michel Haller, SPF, Univ. of Appl. Sciences Eastern Switzerland (OST)



Co-funded by
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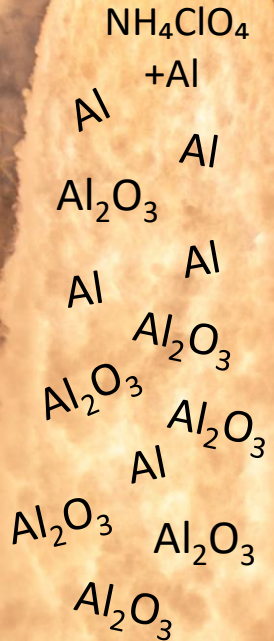
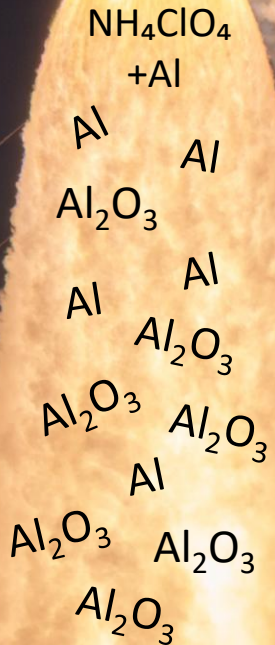
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NASA ARTEMIS II MOON MISSION AND EU REVEAL-STORAGE PROJECT

WHAT DO THEY HAVE IN COMMON?

Aluminium Energy Carrier



OVERVIEW

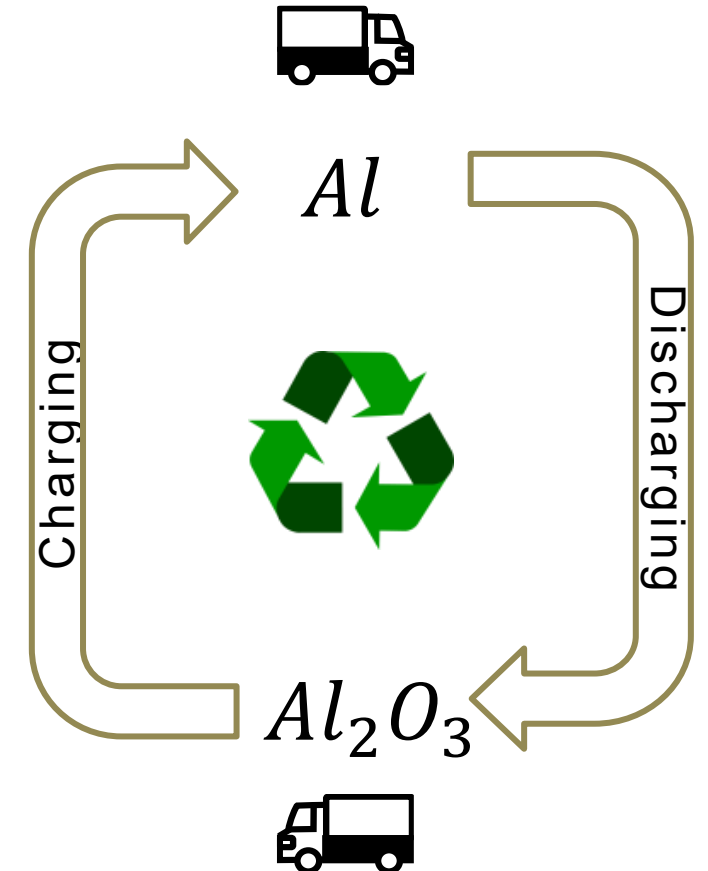
1. Why Aluminium?

2. Power-to-Aluminium (Storage / Fuel Charging)

3. Alu-to-Energy (Storage / Fuel Discharging)

4. Closing the Material Cycle

5. Life Cycle Assessment



1. WHY ALUMINIUM?



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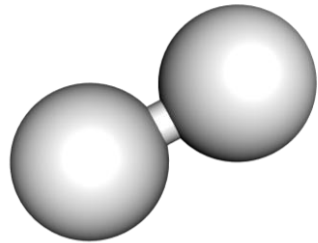
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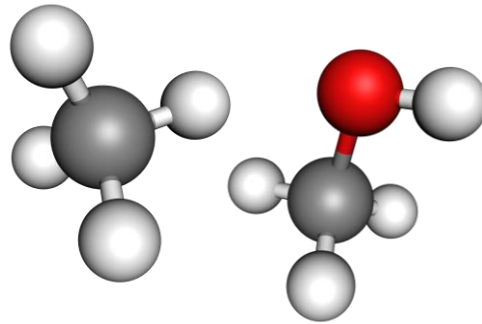
OPTIONS FOR SEASONAL STORAGE OF ENERGY

Hydrogen



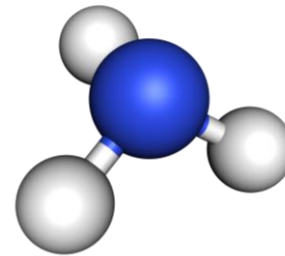
Disadvantage:
vol. density /
storability

Carbohydrates Methane / Methanol



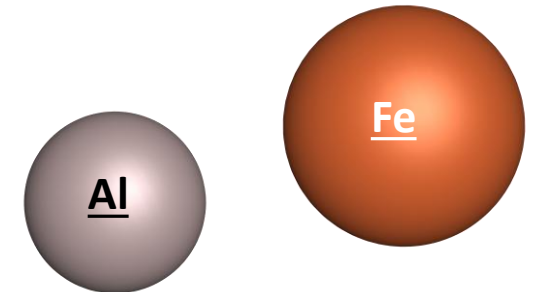
Disadvantage:
carbon:
where from and
where to?

Ammonia



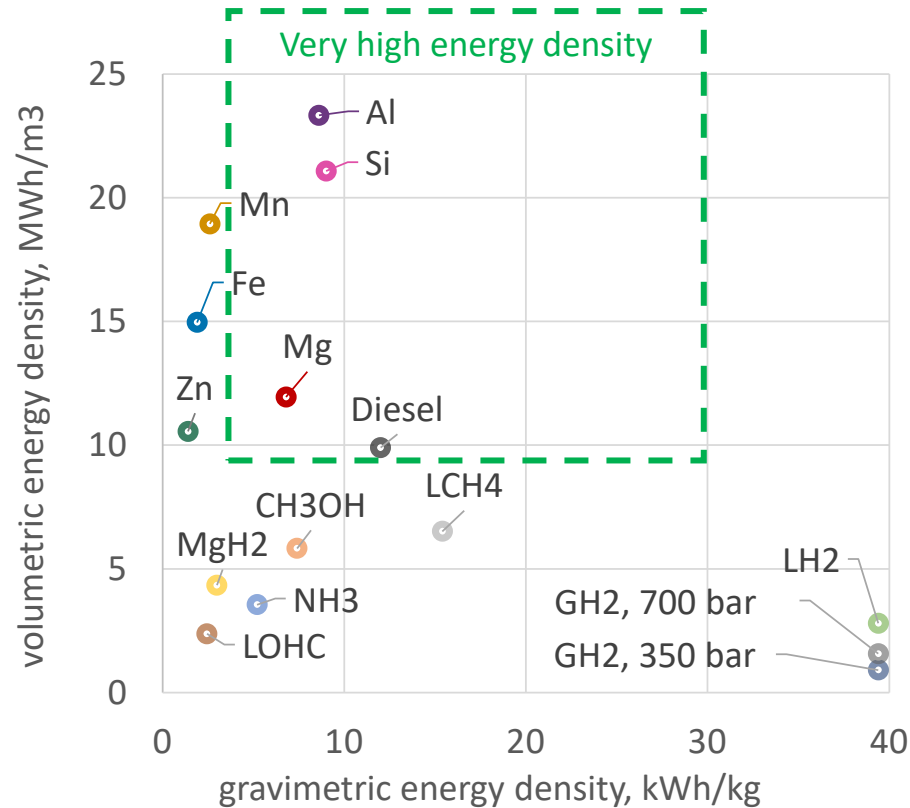
Disadvantage:
toxicity

Metal Energy Carriers: Al, Fe

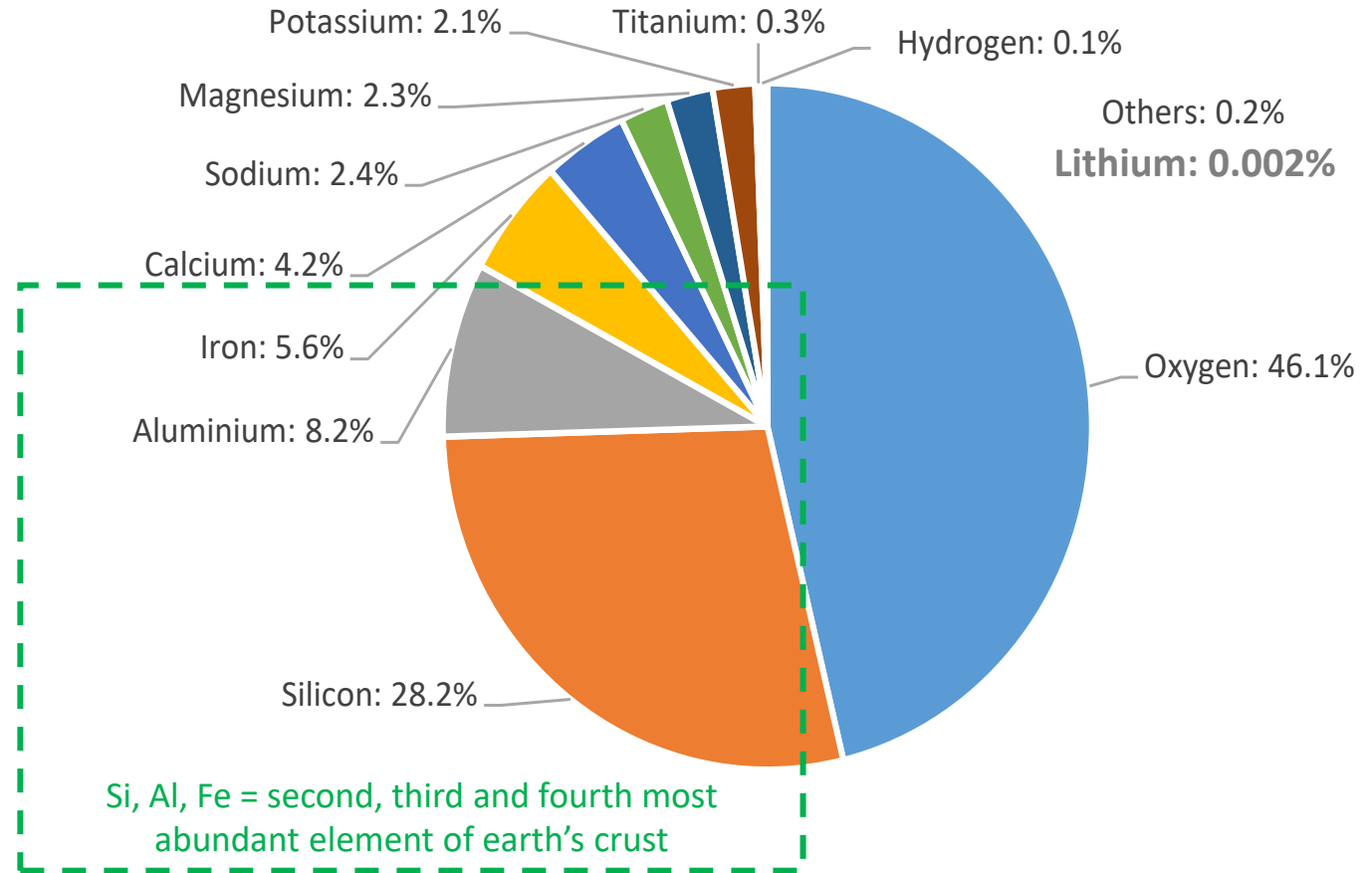


Disadvantage:
technology readiness

ENERGY DENSITY, AVAILABILITY, LOW COST



Elements in the Earth's crust




2. POWER-TO-ALU



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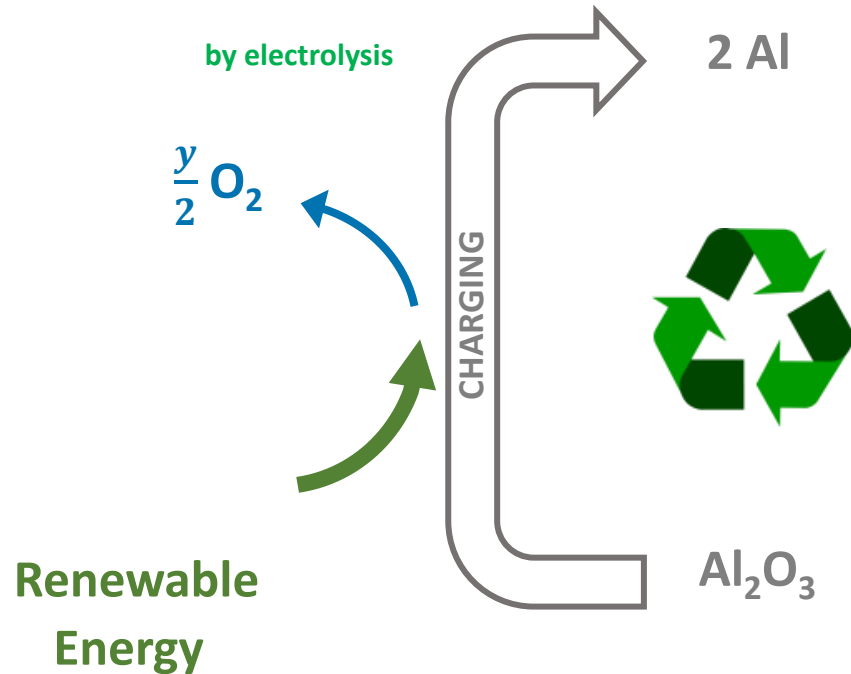
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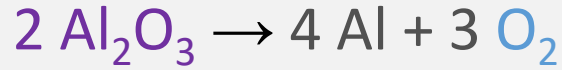
POWER-TO-ALU: REDUCTION OF ALUMINA

Power-to-Metal:



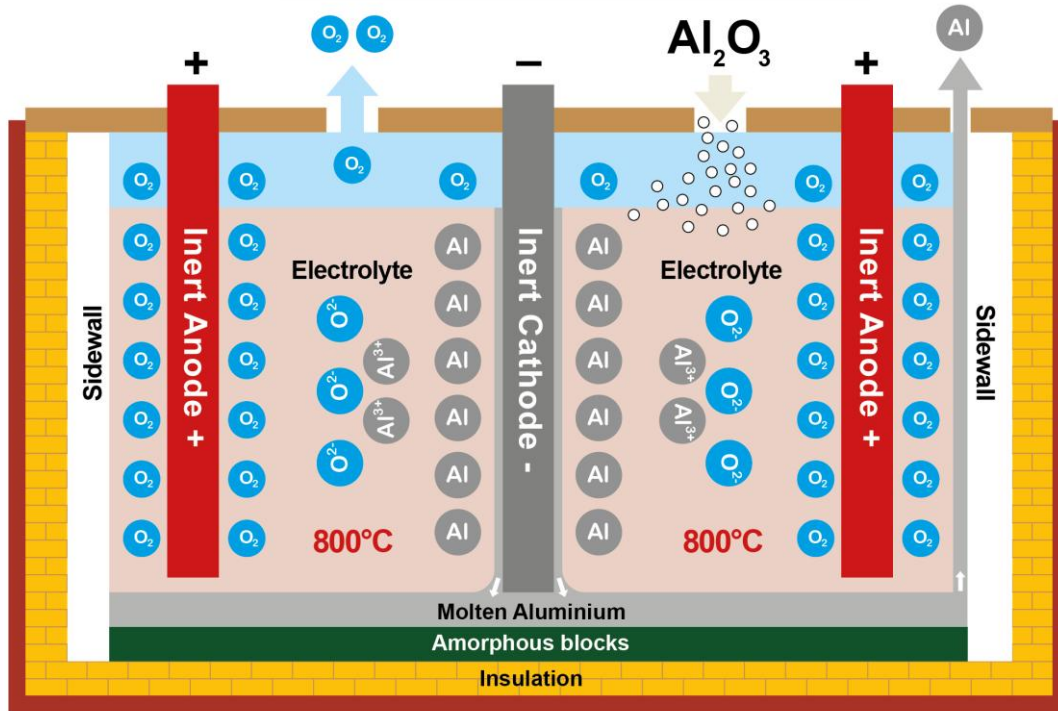
Source: SPF 2024, [PeakMetal](#)

POWER-TO-ALU: INERT ANODE PROCESS



Benefits compared to traditional Hall Héroult process:

- No emissions of CO₂, CF_x, SO₂, cyanides
- 20% less energy demand
- Zero CO₂ compliance costs & taxes
- Modular power feeding during peak hours & power shortage periods for optimal power price
- 50% less space required for same production capacity
- 40% less investment cost
- 30% less operation cost and no carbon anodes



Vertical Inert Anode and Cathodes in low temperature electrolyte

Source: Copyright© Arctus 2025, published with permission




3. ALU-TO-ENERGY



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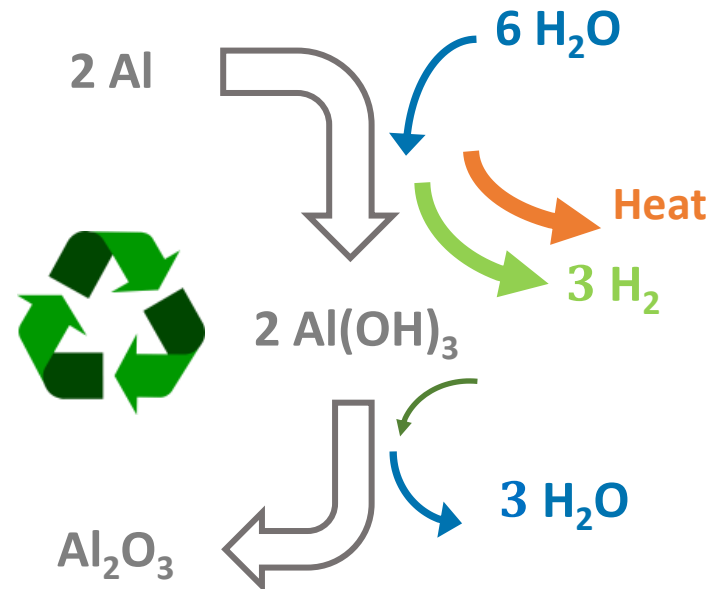
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PATHWAYS FOR ALU-TO-ENERGY

Metal-Water Reaction

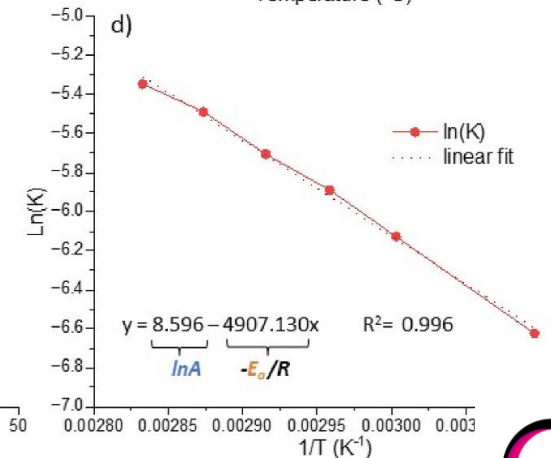
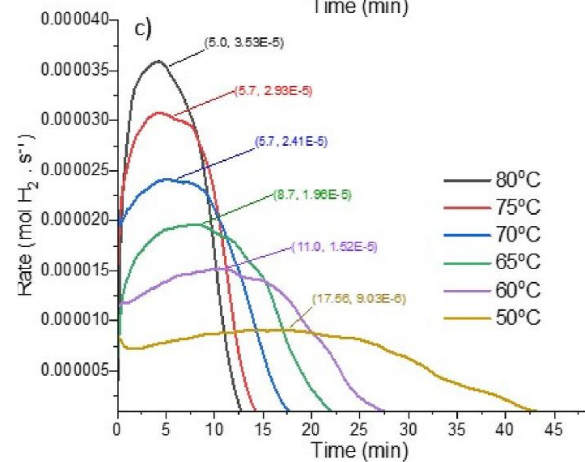
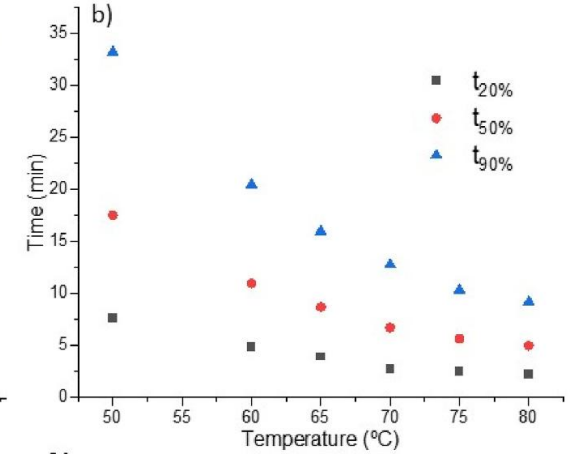
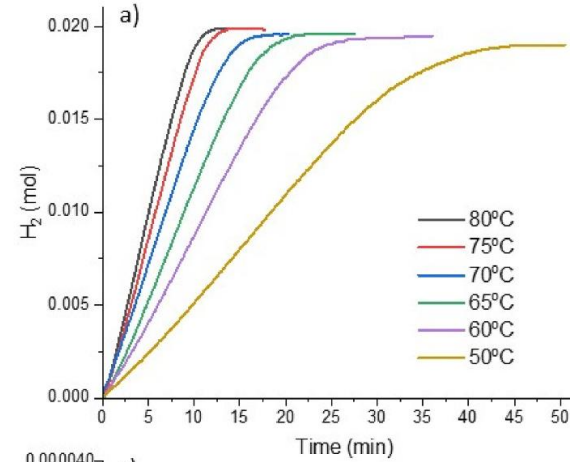
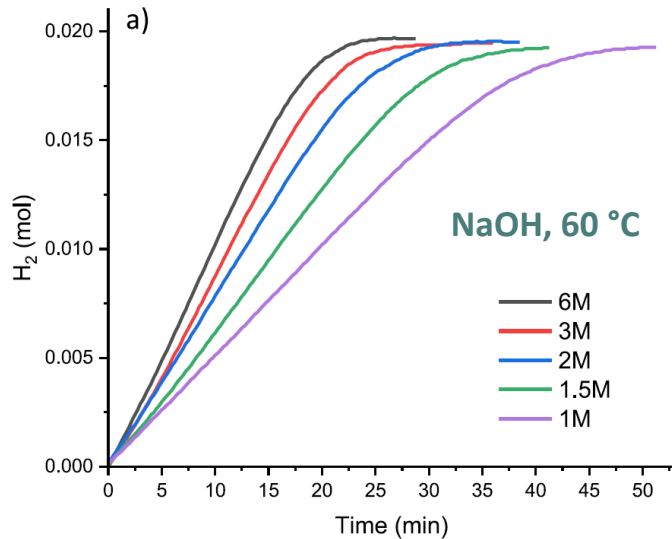
~20 – 150 °C



Reaction with liquid water
(e.g., with sodium hydroxide):
occurs at **ambient to ~150 °C**

ALU-TO-ENERGY: LOW TEMP. WATER

Low temperature aluminium water reactions depend on **alloy, production process, pH and temperature**

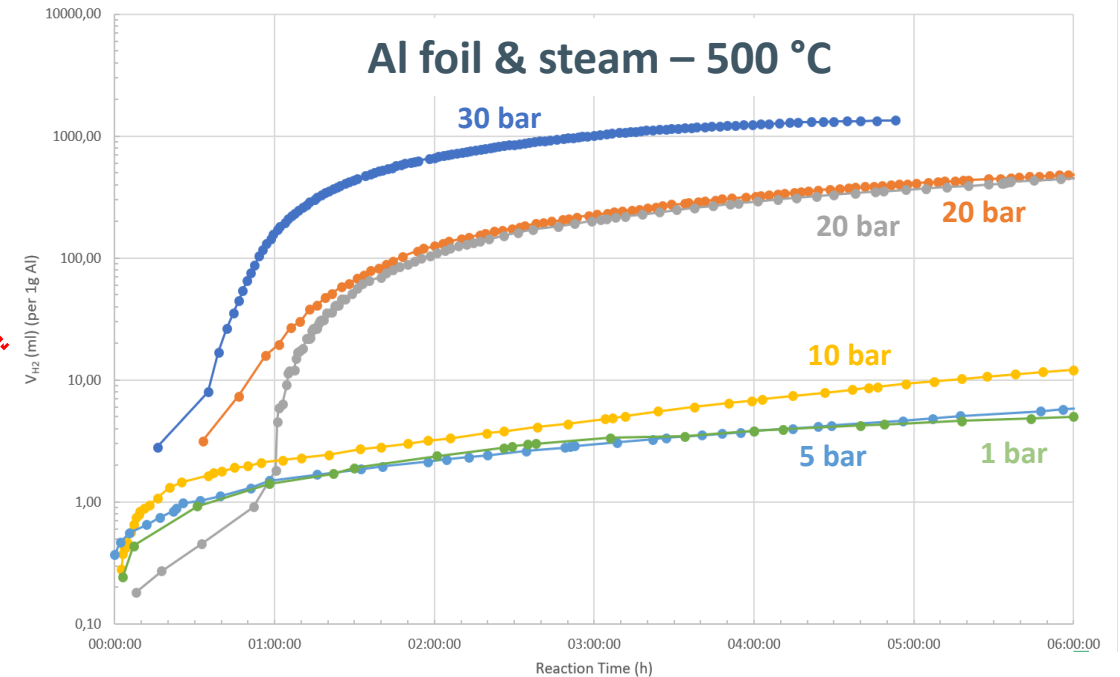
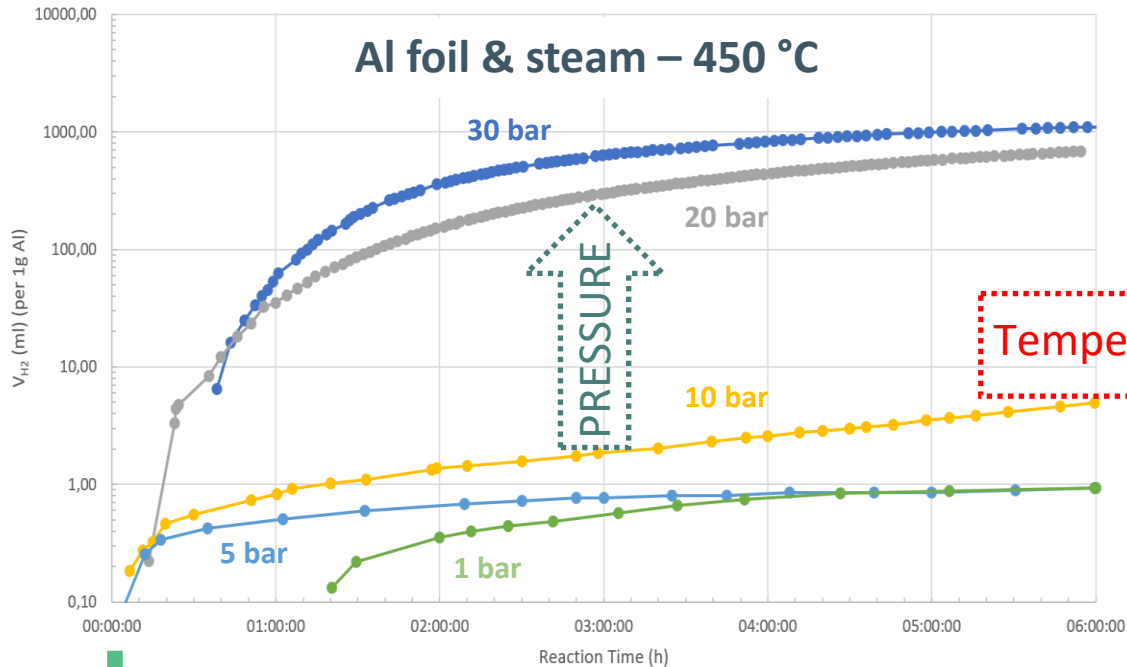


Source: Agote-Arán et al. 2025, [International Journal of Hydrogen Energy](https://doi.org/10.1016/j.ijhe.2025.101010)

Results shown are for **cylindrical Al-6060 cut-wire samples** of around **1 mm** in size.

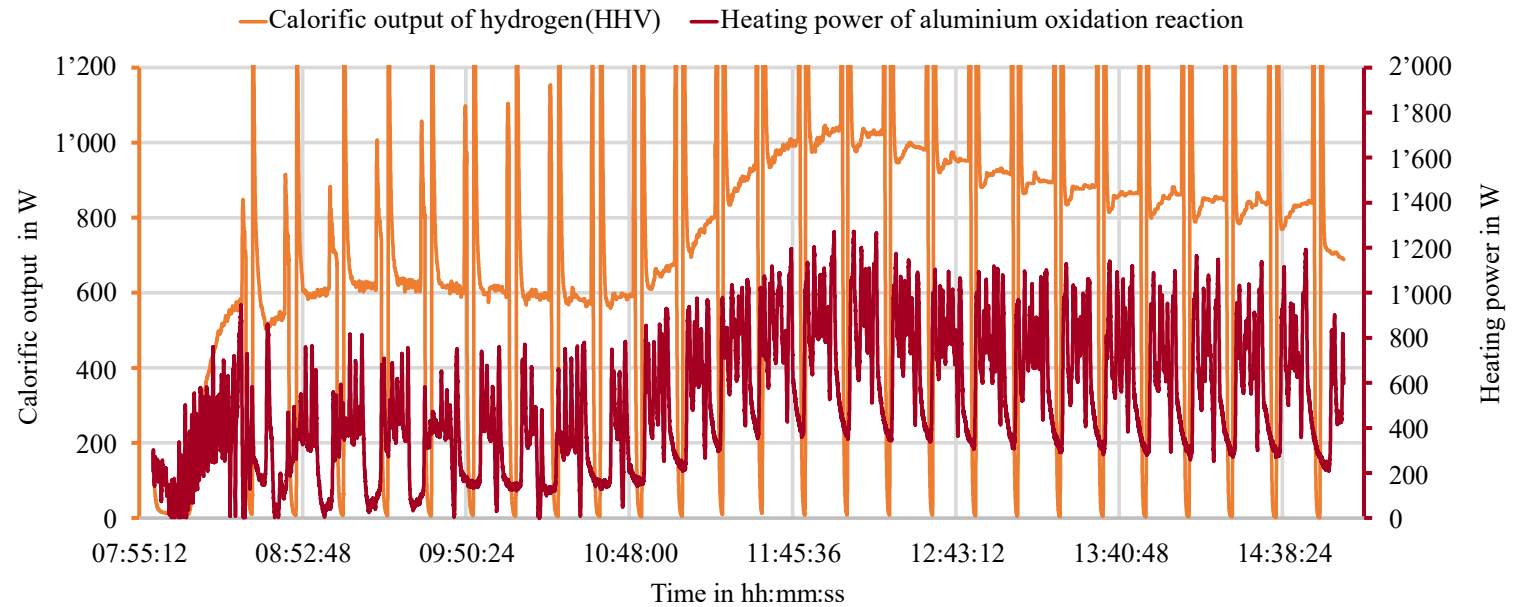
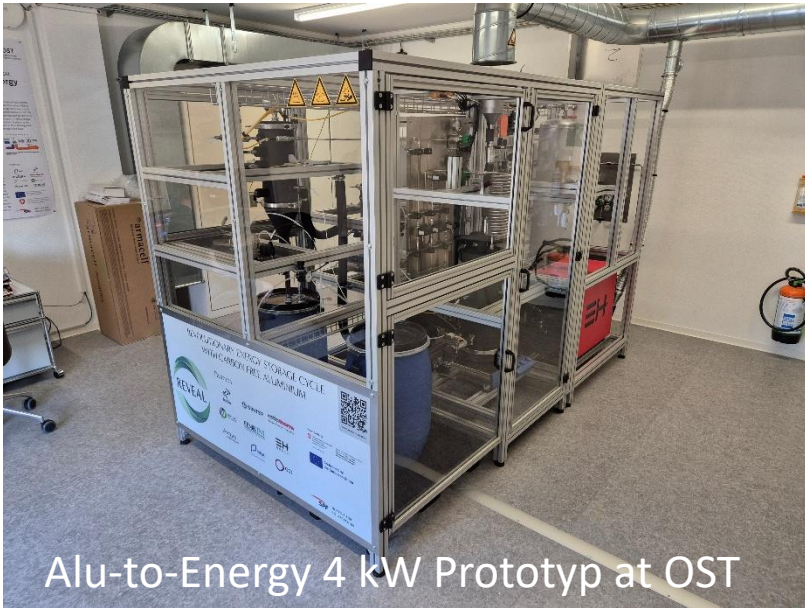
ALU-TO-ENERGY: STEAM

- High temperature aluminium & steam reactions depend on alloy, production process, temperature and pressure
- Shown experiments are with Al-foil, 1 g, 20 μm, for fast energy transformation with this material it needs $\geq 350\text{ °C}$ at 20 bar



Hydrogen yield rises sharply with temperature and pressure, especially from 10 to 30 bar at 450–500 °C.

ALU-TO-ENERGY: 4 kW PROTOTYPE



LIFE CYCLE ASSESSMENT



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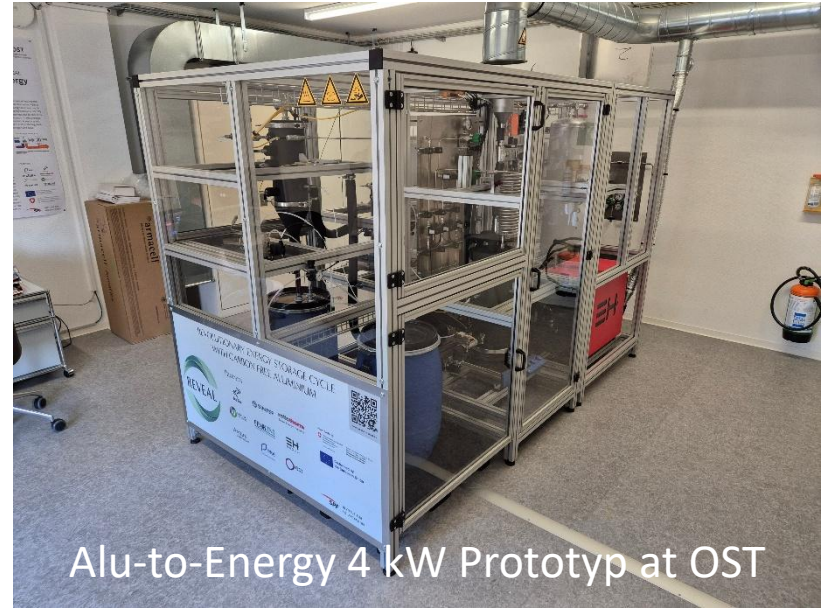
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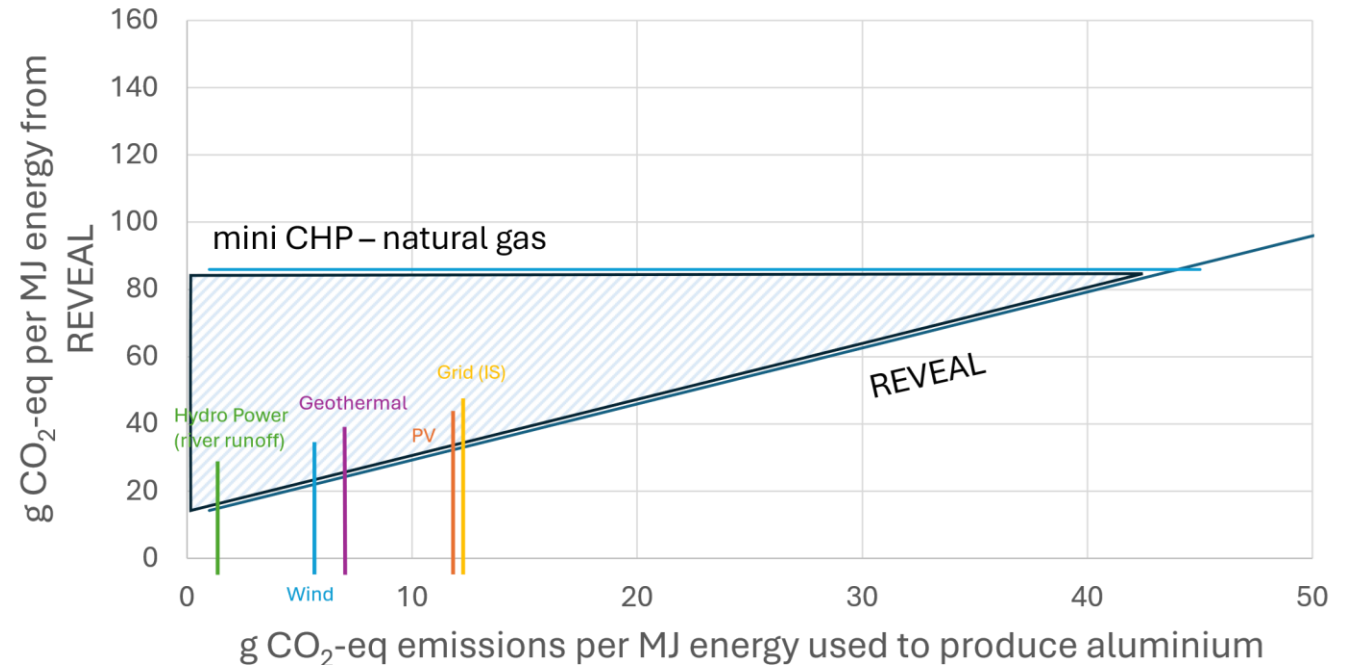
ALU-TO-ENERGY: CLOSED CYCLE



LIFE CYCLE ASSESSMENT

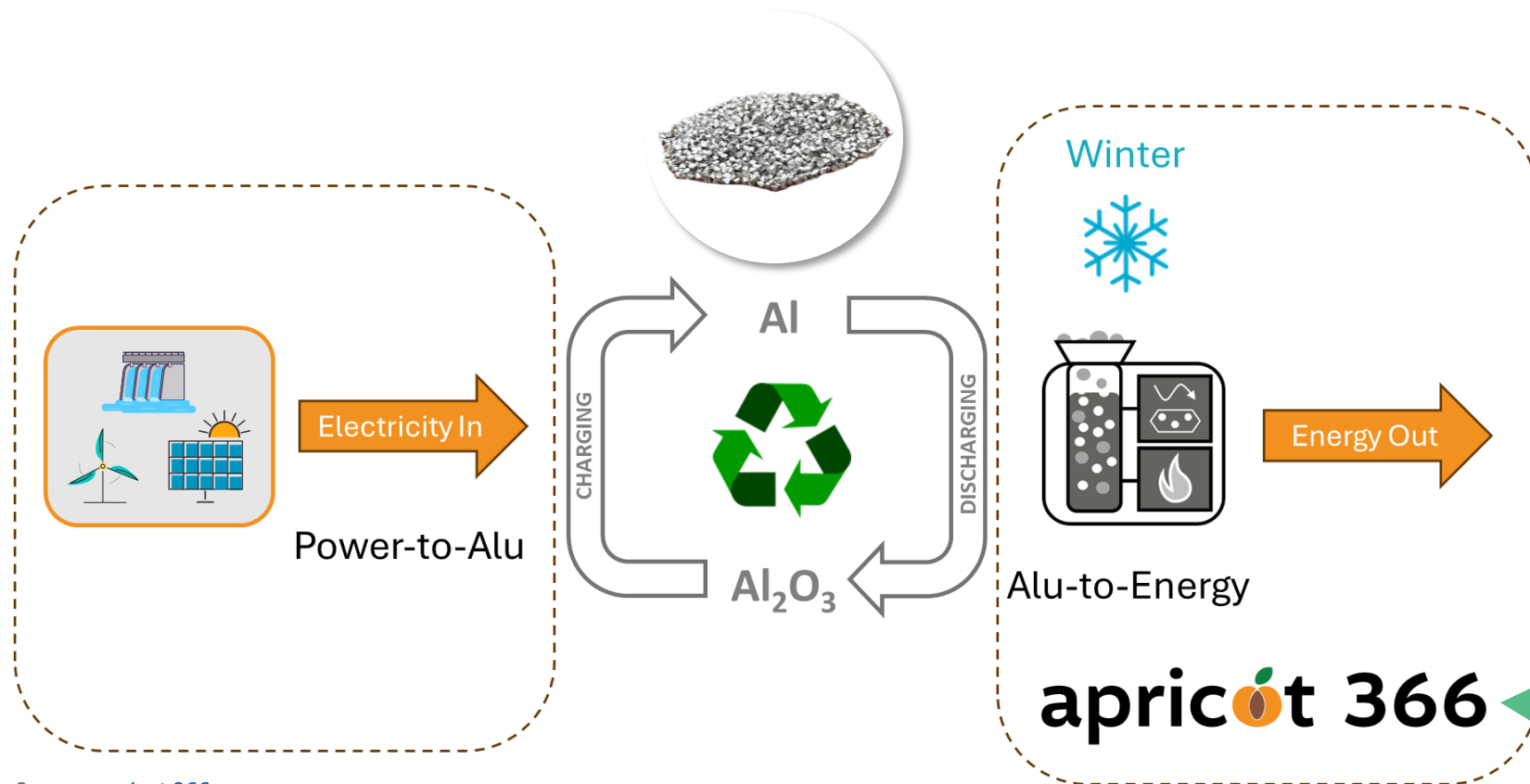
Global Warming Potential: CO₂ free aluminium cycle

- **No direct CO₂ emissions** from the aluminium production.
- **~60% of GHG emissions** from **electricity** input with PV-GWP of 10 g CO_{2eq}/MJ (dominant factor is Al electrolysis)
- **Transport between IS ↔ CH** via cargo ship, electric train, and last-mile diesel trucking accounts for only **~7% of total GWP**
- **Total aluminium cycle GHG emissions** are estimated at **15-34 g CO_{2eq} per MJ**, depending on electricity from wind, hydro or geothermal, grid (IS)



CLOSING THE MATERIAL CYCLE

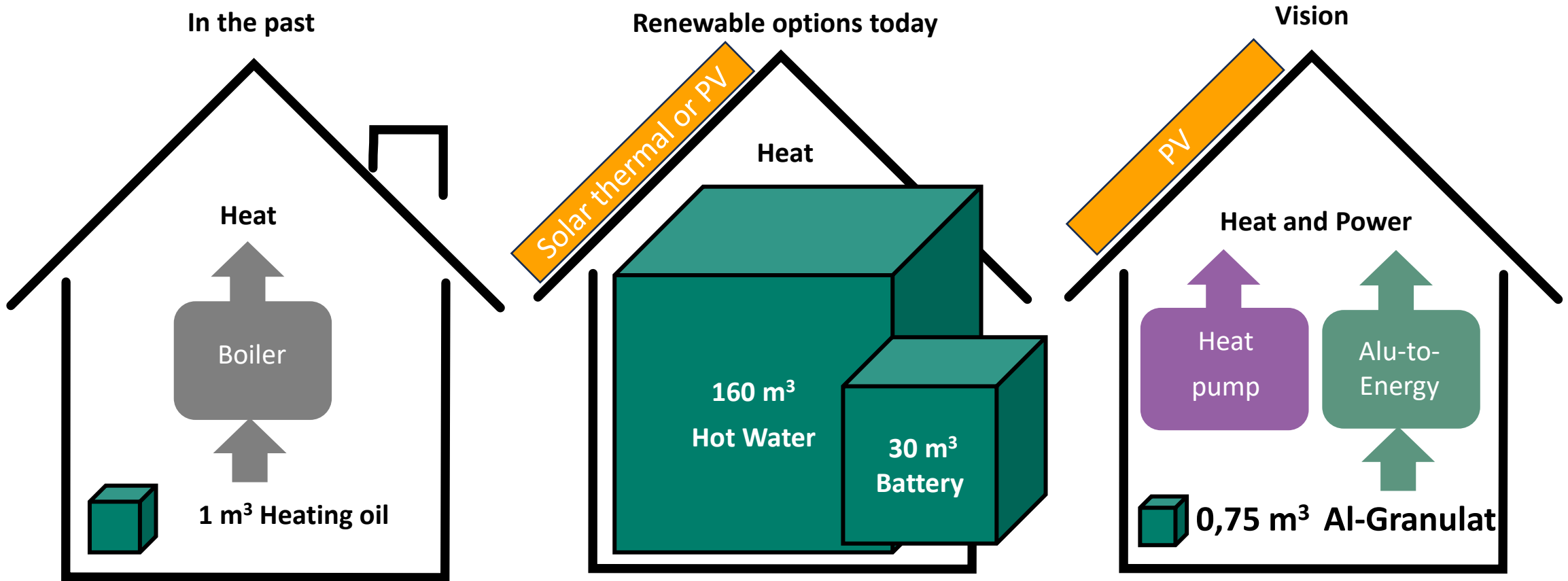
Energy from CO₂-free aluminium: Circular economy



apricot 366 is a start-up out of the REVEAL project, which uses aluminium from the inert anode process for producing hydrogen and heat.

Source: [apricot 366](https://www.apricot366.com)

APPLICATION FOR HEAT AND ELECTRICITY



Storage volume requirement to provide a winter heat demand of **11 000 kWh.**

Source: OST-SPF

THANK YOU!



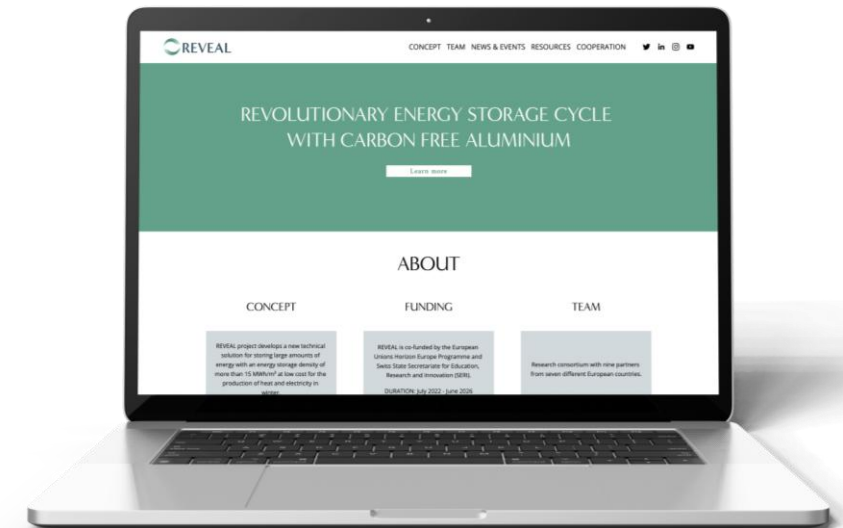
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